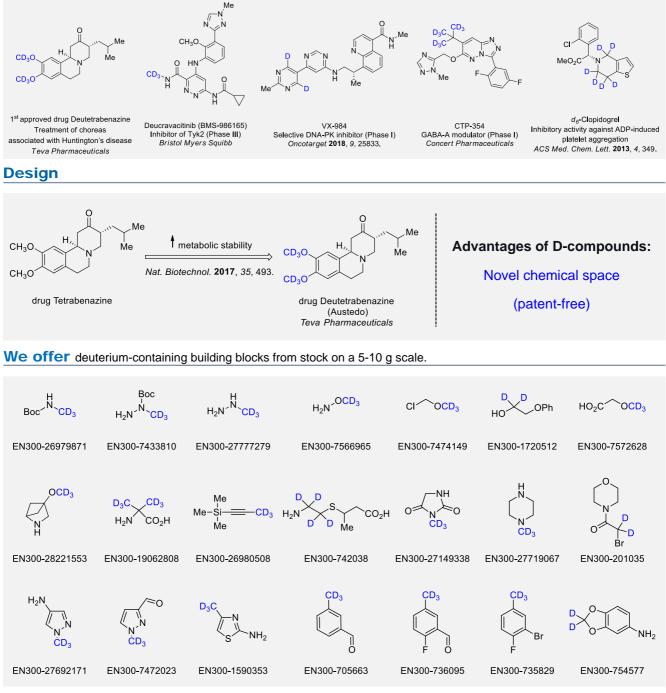
Introduction

Selective incorporation of deuterium into organic molecules can profoundly alter their physical, chemical and biological properties. Compared to hydrogen, deuterium displays a smaller molar volume (by 0.140 cm³/mol per atom), is less lipophilic (Δ logPoct = - 0.006) and might display a slightly different pKa. More importantly, C-D bonds are shorter (by 0.005 Å) and at times more stable to oxidative processes. This helps protect deuterated compounds from enzymatic degradation in the liver, increasing the compound's half-life.¹⁻⁴ In this context, Enamine offers a library of deuterium-containing building blocks for drug design.



References

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